

Lesson 2: Semiconductor Basics

What is a semiconductor?

(Razavi 2.1.1 ~ 2.1.2

Youtube: Lec. 1, 21 min. to End

Lec. 2, Beginning to 34 min.)

	IIIA	IVA	VA	VIA	
	5 B	6 C	7 N	8 O	
	13 Al	14 Si	15 P	16 S	
IIB	30 Zn	31 Ga	32 Ge	33 As	34 Se
	48 Cd	49 In	50 Sn	51 Sb	52 Te

Group IV elements (for example, Si) has four valence electrons

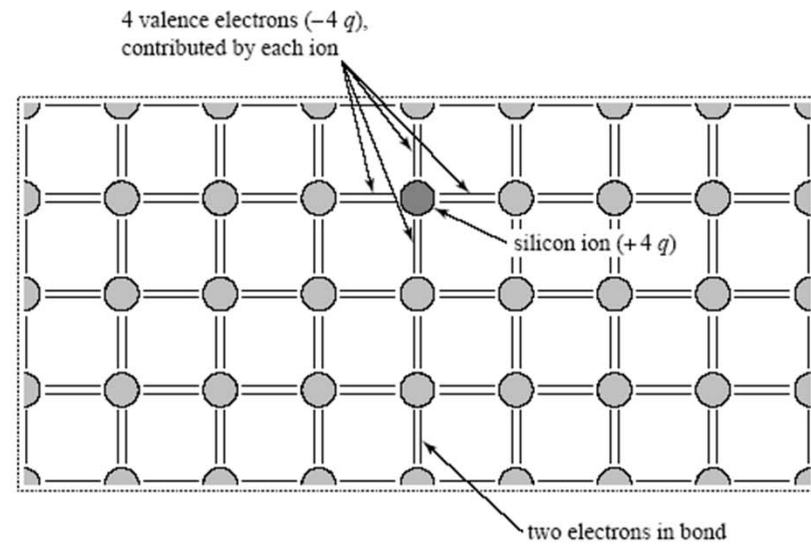
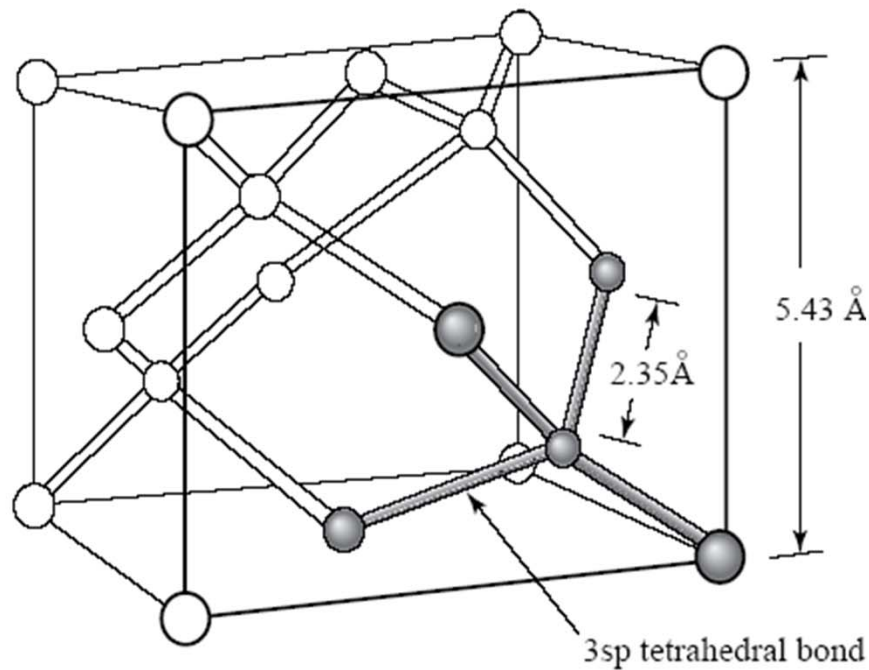
(Also for combination of

Group III and Group V elements, for example, GaAs, and

Group II and Group VI elements, for example, ZnSe)

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How does it look? Diamond-like crystal structure



2-dimensional model

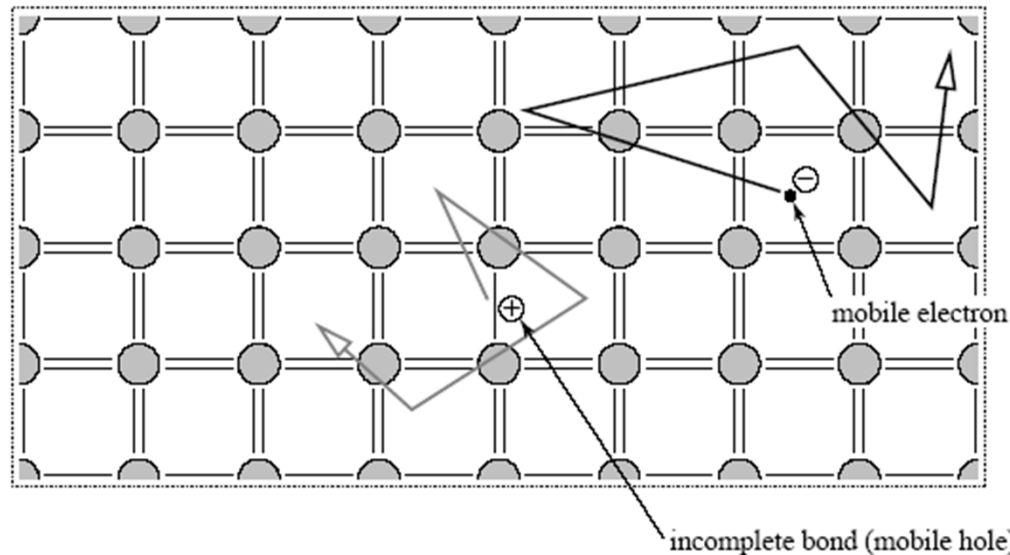
Can currents flow?

Si atoms share electrons with neighboring atoms

→ each atom can have 8 electrons

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At finite temperature, some electrons escape from bonds creating mobile electrons and holes



$$n \text{ (density of electrons)} = p \text{ (density of holes)} = 5.2 \times 10^{15} \times T^{2/3} \exp(-E_g/2kT)$$

➔ Current conduction due to electrons and holes

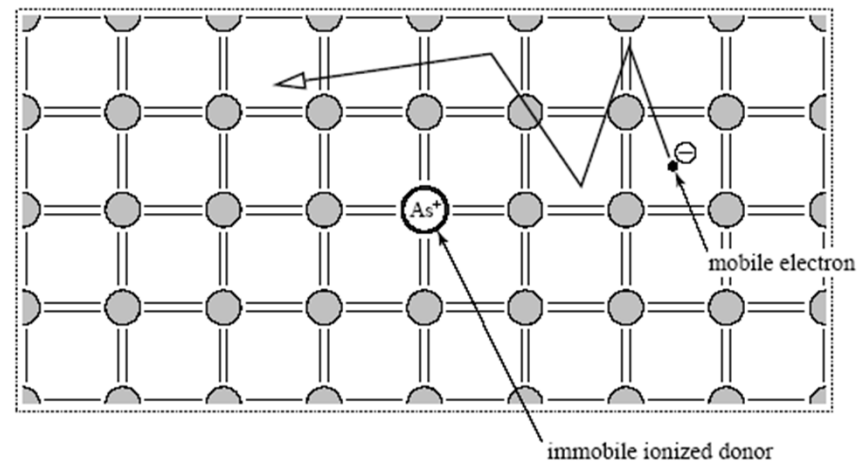
n_i at RT for Si : about 10^{10} (1/cm³) No. of Si atoms: 5×10^{22} (1/cm³)

$n \times p = n_i^2$ **Intrinsic semiconductor**

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What happens when Group V (donor) atoms are added (doping)?

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	5 B	6 C	7 N	8 O
	13 Al	14 Si	15 P	16 S
IIB	30 Zn	31 Ga	32 Ge	33 As
	48 Cd	49 In	50 Sn	51 Sb
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Extra electrons!

If $N_d \gg n_i$, $n \sim N_d$ (majority carrier)

Extrinsic semiconductor: N-type

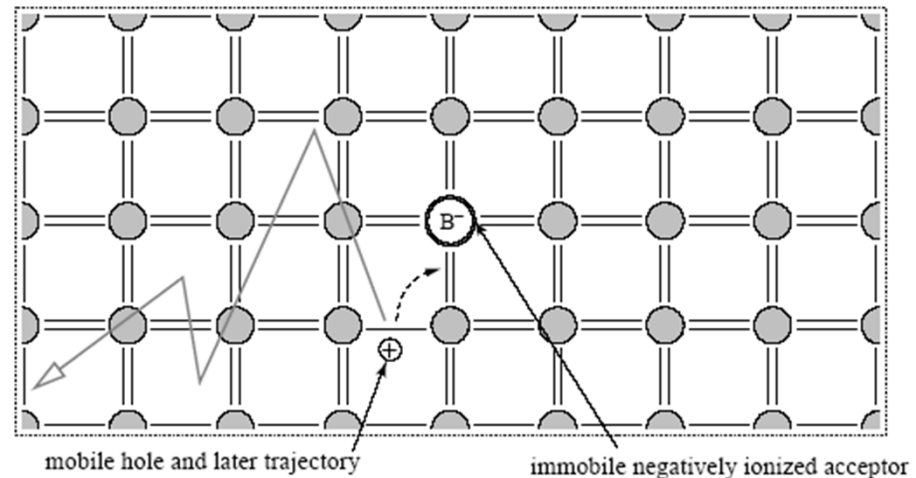
$p \sim n_i^2 / N_d$ (minority carrier)

→ More conductive (less resistive)

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What happens when Group III (acceptor) atoms are added (doping)?

	IIIA	IVA	VA	VIA
	5 B	6 C	7 N	8 O
	13 Al	14 Si	15 P	16 S
IIB	30 Zn	31 Ga	32 Ge	33 As
	48 Cd	49 In	50 Sn	51 Sb
			52 Te	



Extra holes!

If $N_a \gg n_i$, $p \sim N_a$ (majority carrier)

$n \sim n_i^2 / N_a$ (minority carrier)

Extrinsic semiconductor: P-type

→ More conductive (less resistive)

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For Thursday (9/3)

Razavi Youtube lectures:

Lec. 1, 21 min. to End, Lec. 2, Beginning to 34 min.

→ Quiz 1

Quiz will start at 9 o'clock. Don't be late.