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# The hydrogen atom

- I. Solving the hydrogen atom problem
- II. Informal solution for the relative motion

# Multiple Particle Systems

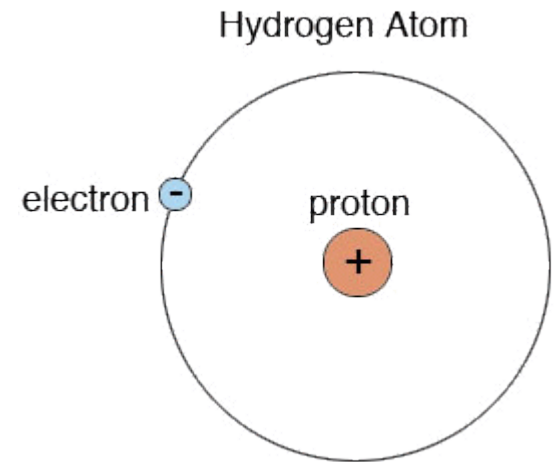
## Schrödinger's equation

$$\hat{H}\psi = E\psi$$

State of the entire system

Hamiltonian operator for the entire system

Total energy of the entire system



- ✓ Electron coordinates:  $x_e, y_e, z_e$
  - ✓ Proton coordinates:  $x_p, y_p, z_p$
- } 6 coordinates for the entire system

$$\left[ -\frac{\hbar^2}{2m_e} \nabla_e^2 - \frac{\hbar^2}{2m_p} \nabla_p^2 + V(|\mathbf{r}_e - \mathbf{r}_p|) \right] \psi(x_e, y_e, z_e, x_p, y_p, z_p)$$

$$V(|\mathbf{r}_e - \mathbf{r}_p|) = -\frac{e^2}{4\pi\epsilon_0 |\mathbf{r}_e - \mathbf{r}_p|} = E\psi(x_e, y_e, z_e, x_p, y_p, z_p)$$

# Simplification of Coordinates

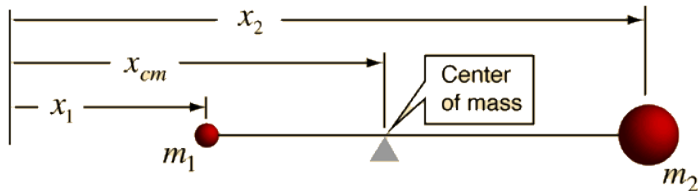
## 1. Relative positions coordinates

$$x = x_e - x_p \quad y = y_e - y_p \quad z = z_e - z_p$$

Relative vector:  $\mathbf{r} = x\mathbf{i} + y\mathbf{j} + z\mathbf{k}$

$$r = \sqrt{x^2 + y^2 + z^2} = |\mathbf{r}_e - \mathbf{r}_p|$$

## 2. Center of mass coordinates



$$\mathbf{R} = \frac{m_e \mathbf{r}_e + m_p \mathbf{r}_p}{M}$$

$$M = m_e + m_p$$

$$= X\mathbf{i} + Y\mathbf{j} + Z\mathbf{k}$$

$$X = \frac{m_e x_e + m_p x_p}{M}$$

Similar way to Y and Z

# Second Derivatives Of Coordinates

Schrödinger's equation

$$\left[ -\frac{\hbar^2}{2m_e} \nabla_e^2 - \frac{\hbar^2}{2m_p} \nabla_p^2 + V(|\mathbf{r}_e - \mathbf{r}_p|) \right] \psi(x_e, y_e, z_e, x_p, y_p, z_p) = E\psi(x_e, y_e, z_e, x_p, y_p, z_p)$$

Should be substituted!

$$\begin{aligned} \left. \frac{\partial^2}{\partial x_e^2} \right|_{x_p} &= \left. \frac{\partial}{\partial x_e} \right|_{x_p} \left( \left. \frac{\partial}{\partial x_e} \right|_{x_p} \right) = \frac{m_e}{M} \left. \frac{\partial}{\partial x_e} \right|_{x_p} \left. \frac{\partial}{\partial X} \right|_x + \left. \frac{\partial}{\partial x_e} \right|_{x_p} \left. \frac{\partial}{\partial x} \right|_x \\ &= \left( \frac{m_e}{M} \right)^2 \left. \frac{\partial^2}{\partial X^2} \right|_x + \left. \frac{\partial^2}{\partial x^2} \right|_x + \frac{m_e}{M} \left( \left. \frac{\partial}{\partial x} \right|_x \left. \frac{\partial}{\partial X} \right|_x + \left. \frac{\partial}{\partial X} \right|_x \left. \frac{\partial}{\partial x} \right|_x \right) \end{aligned}$$

Similarly,

$$\left. \frac{\partial^2}{\partial x_p^2} \right|_{x_e} = \left( \frac{m_p}{M} \right)^2 \left. \frac{\partial^2}{\partial X^2} \right|_x + \left. \frac{\partial^2}{\partial x^2} \right|_x - \frac{m_p}{M} \left( \left. \frac{\partial}{\partial x} \right|_x \left. \frac{\partial}{\partial X} \right|_x + \left. \frac{\partial}{\partial X} \right|_x \left. \frac{\partial}{\partial x} \right|_x \right)$$

# Hamiltonian Operator With New Coordinates

$$\begin{aligned}\frac{1}{m_e} \frac{\partial^2}{\partial x_e^2} + \frac{1}{m_p} \frac{\partial^2}{\partial x_p^2} &= \frac{m_e + m_h}{M^2} \frac{\partial^2}{\partial X^2} + \left( \frac{1}{m_e} + \frac{1}{m_p} \right) \frac{\partial^2}{\partial x^2} \\ &= \frac{1}{M} \frac{\partial^2}{\partial X^2} + \frac{1}{\mu} \frac{\partial^2}{\partial x^2}\end{aligned}$$

$$\mu = \frac{m_e m_p}{m_e + m_p}$$

Reduced mass



$$\hat{H} = -\frac{\hbar^2}{2M} \nabla_{\mathbf{R}}^2 - \frac{\hbar^2}{2\mu} \nabla_{\mathbf{r}}^2 + V(\mathbf{r})$$

$$\nabla_{\mathbf{R}}^2 \equiv \frac{\partial^2}{\partial X^2} + \frac{\partial^2}{\partial Y^2} + \frac{\partial^2}{\partial Z^2}$$

$$\nabla_{\mathbf{r}}^2 \equiv \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$$

# Solving Schrödinger's Equation

Wave equation can be written as  $\psi(\mathbf{R}, \mathbf{r}) = S(\mathbf{R})U(\mathbf{r})$

Schrödinger's equation

$$-U(\mathbf{r})\frac{\hbar^2}{2M}\nabla_{\mathbf{R}}^2 S(\mathbf{R}) + S(\mathbf{R})\left[-\frac{\hbar^2}{2\mu}\nabla_{\mathbf{r}}^2 + V(\mathbf{r})\right]U(\mathbf{r}) = ES(\mathbf{R})U(\mathbf{r})$$



$$-\frac{1}{S(\mathbf{R})}\frac{\hbar^2}{2M}\nabla_{\mathbf{R}}^2 S(\mathbf{R}) = E - \frac{1}{U(\mathbf{r})}\left[-\frac{\hbar^2}{2\mu}\nabla_{\mathbf{r}}^2 + V(\mathbf{r})\right]U(\mathbf{r}) = \underline{E_{CoM}}$$

Constant

# Solution For Two Particles

$$-\frac{\hbar^2}{2M} \nabla_{\mathbf{R}}^2 S(\mathbf{R}) = E_{CoM} S(\mathbf{R}) \quad \text{Center of mass motion}$$

Similar to single particle  $S(\mathbf{R}) = \exp(i\mathbf{K} \cdot \mathbf{R}) \quad E_{CoM} = \frac{\hbar^2 K^2}{2M}$

$$\left[ -\frac{\hbar^2}{2\mu} \nabla_{\mathbf{r}}^2 + V(\mathbf{r}) \right] U(\mathbf{r}) = E_H U(\mathbf{r}) \quad \text{Relative motion}$$

$$E_H = E - E_{CoM}$$

Corresponds to the “internal” relative motion of the electron and proton

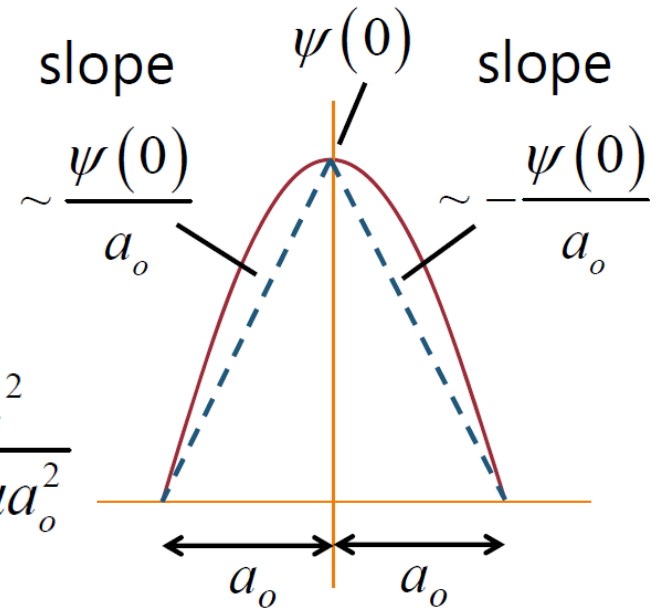
# Rough Energy Calculation of Relative Motion

## Potential energy

$$\langle E_{potential} \rangle \approx -\frac{e^2}{4\pi\epsilon_0 a_0} \quad a_0: \text{Bohr radius}$$

## Kinetic energy

$$\left. \begin{aligned} \text{Kinetic energy operator: } & -(\hbar^2 / 2\mu)\nabla^2 \\ \nabla^2 \sim & \frac{[-\psi(0)/a_0] - [\psi(0)/a_0]}{2a_0} \\ & \sim -\psi(0)/a_0^2 \end{aligned} \right\} \langle E_{kinetic} \rangle \approx \frac{\hbar^2}{2\mu a_0^2}$$



## Total energy

$$\langle E_{total} \rangle = \langle E_{kinetic} \rangle + \langle E_{potential} \rangle \approx \frac{\hbar^2}{2\mu a_0^2} - \frac{e^2}{4\pi\epsilon_0 a_0}$$



# Bohr Radius and Rydberg Energy

$$\langle E_{total} \rangle = \langle E_{kinetic} \rangle + \langle E_{potential} \rangle \approx \frac{\hbar^2}{2\mu a_0^2} - \frac{e^2}{4\pi\epsilon_0 a_0}$$

Minimization of energy with  $a_0 = \frac{4\pi\epsilon_0 \hbar^2}{e^2 \mu} \sim 0.529 \text{ \AA}$

Minimized energy  $\langle E_{total} \rangle = -\frac{\hbar^2}{2\mu a_0^2} = -\frac{\mu}{2} \left( \frac{e^2}{4\pi\epsilon_0 \hbar} \right)^2 \sim -\frac{13.6 eV}{\text{Rydberg Energy}}$